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(21) International Application Number: PCT/US96/10341 (22) International Filing Date: 13 June 1996 (13.06.96) (30) Priority Data: 08/515,187 11 August 1995 (11.08.95) US (71) Applicant: MINNESOTA MINING AND MANUFACTURING COMPANY [US/US]; 3M Center, P.O. Box 33427, Saint Paul, MN 55133-3427 (US). (72) Inventor: DEVINY, E., John; P.O. Box 33427, Saint Paul, MN 55133-3427 (US). (74) Agents: O'SHAUGHNESSY, Christine, T. et al.; Minnesota Mining and Manufacturing Company, Office of Intellectual Property Counsel, P.O. Box 33427, Saint Paul, MN 55133-3427 (US).	(81) Designated States: AL, AM, AT, AU, AZ, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IL, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, UZ, VN, ARIPO patent (KE, LS, MW, SD, SZ, UG), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i>	
(54) Title: INITIATOR SYSTEM AND ADHESIVE COMPOSITION MADE THEREWITH (57) Abstract Systems for initiating the polymerization of acrylic monomers comprising (1) organoborane amine complexes and (2) bireactive decomplexers preferably comprising at least one free-radically polymerizable group and at least one amine-reactive group in the same molecule. The decomplexer is capable of forming a covalent bond with both the acrylic monomers and amine complex, resulting in a reduced level of mobile constituents. Furthermore, when the organoborane amine complex of the initiator system comprises a polyamine compound, polymerized acrylic compositions having improved solvent resistance are advantageously provided.		

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INITIATOR SYSTEM AND ADHESIVE COMPOSITION MADE THEREWITH

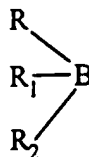
Field of the Invention

This invention relates generally to systems capable of initiating the polymerization of acrylic monomers. More specifically, the invention relates to those initiator systems comprising (1) organoborane amine complexes and (2) bireactive decomplexers. The invention further relates to the use of these systems for initiating the polymerization of acrylic monomers, as well as acrylic adhesive compositions made therewith. The adhesive compositions have good adhesion to low energy surfaces and excellent solvent resistance.

Description of the Related Art

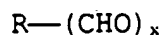
Systems for initiating the polymerization of acrylic monomers to make adhesive compositions are known in the art. U.S. Patent Nos. 5,106,928, 5,286,821, and 5,310,835, for example, disclose a two part initiator system that is reportedly useful in acrylic adhesive compositions. The first part of this two part system includes a stable organoborane amine complex and the second part includes an activator. The activator liberates the organoborane compound by removing the amine group, thereby allowing the organoborane compound to initiate the polymerization process.

The organoborane compound of the complex has the general formula:



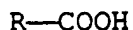
where R, R₁ and R₂ are either alkyl groups having 1 to 10 carbon atoms or phenyl groups. Useful amine compounds for the complex include n-octylamine, 1,6-diaminohexane, diethylamine, dibutylamine, diethylenetriamine, dipropylenediamine, ammonia, 1,3-propylenediamine, and 1,2-propylenediamine.

U.S. Patent No. 5,286,821 reports that suitable activators for liberating the organoborane compound include aldehydes having the general structure:



wherein R is alkyl of 1 to 10 carbon atoms or aryl having 6 to 10 carbon atoms and x is 1 to 2. Examples include benzaldehyde, o-, m-, p-nitrobenzaldehyde, 2,4-dichlorobenzaldehyde, p-tolylaldehyde, and 3-methoxy-4-hydroxybenzaldehyde.

U.S. Patent 5,310,835 reports that suitable activators also include organic acids having the structure:



wherein R is H, alkyl, or alkenyl having 1 to 8 carbon atoms. Examples include acrylic acid, methacrylic acid, benzoic acid, and p-methoxybenzoic acid.

Fujisawa, Imai, and Mashuhara also describe a system for initiating the polymerization of methyl methacrylate. See Reports of the Institute for Medical and Dental Engineering, 3, 64 (1969). The system comprises a trialkylborane amine complex and an activator such as the chloride of methacrylic or n-butane sulfonic acid, terephthalic acid chloride, benzoyl chloride, p-toluene sulfonic acid chloride, benzene sulfonic acid chloride, methane sulfonic acid chloride, toluene diisocyanate, adipic acid chloride, o-tolyl isocyanate, acetyl chloride, and anhydrous

acetic acid. The initiator system is reportedly useful in providing fast curing resins for dental applications.

Activators (sometimes also referred to as decomplexers) liberate the organoborane compound by bonding (either covalently or ionically) to the amine to form an activator-amine constituent. Most activator-amine constituents remain in the adhesive composition as a mobile constituent that is not incorporated into the polymerized adhesive per se. In general, mobile constituents in adhesive compositions may cause performance problems when, for example, they migrate to the surface of the adhesive thereby disrupting the bond interface. Mobile constituents are also susceptible to attack by solvents, thereby making the adhesive composition less suitable for applications where exposure to solvents is unavoidable.

A need therefore exists for an initiator system that provides an acrylic adhesive composition having a reduced level of mobile constituents and increased solvent resistance. Such an adhesive is less susceptible to performance problems due to migration of mobile constituents to the bond interface and can also be used in various manufacturing applications where adhesives are exposed to organic vapors, oils, fuels, and other solvents such as methylethylketone, tetrahydrofuran, and the like.

SUMMARY OF THE INVENTION

This invention provides systems capable of initiating the polymerization of acrylic monomers. More specifically, the invention provides initiator systems comprising (1) an organoborane amine complex and (2) a bireactive decomplexer. The term "decomplexer" means a compound capable of liberating the organoborane compound by bonding with the amine group, thereby allowing the organoborane to initiate

the polymerization process. Decomplexers are also sometimes referred to as "activators" or "liberators." As used herein, each of these terms refers to a compound capable of liberating the organoborane compound by bonding with the amine group. The term "bireactive" means that the decomplexer is capable of forming a covalent bond with both acrylic monomers and amine portion of the organoborane amine complex. Accordingly, the decomplexer is capable of covalently bonding to the liberated amine in the system and also reacting into the adhesive per se. This results in a reduced level of mobile constituents (i.e., those materials capable of migrating or diffusing through the cured adhesive). Furthermore, when the organoborane amine complex of the initiator system comprises a polyamine compound, polymerized acrylic compositions having improved solvent resistance are advantageously provided.

The decomplexer of the initiator system of the invention preferably comprises at least one free radically polymerizable group (for example, substituted alkenes) and at least one amine-reactive group in the same molecule. Preferably, each group is terminal. More preferably, the decomplexer comprises an acrylic group and an isocyanate group in the same molecule.

Preferred decomplexers include the reaction products of hydroxylated (meth)acrylates or adducts thereof and polyisocyanates or adducts thereof. The term "adduct" means addition-reaction product. The term "polyisocyanate" means a compound having at least two isocyanate groups. The term "(meth)acrylate" includes both acrylate and methacrylate. Preferred adducts of hydroxylated (meth)acrylates include hydroxyacrylate polyesters formed from lactones. Preferred adducts of polyisocyanates include those

formed from active hydrogen containing compounds such as polyols, polythiols, polyamines and the like.

The system of the invention can be used to initiate the polymerization of acrylic monomer(s). Consequently, the invention also relates to a polymerizable acrylic composition that comprises at least one acrylic monomer, an effective amount of an organoborane amine complex, and an effective amount of the decomplexer of the invention for liberating the organoborane to initiate polymerization of the at least one acrylic monomer.

A wide variety of acrylic monomers may be used but those which are preferred include monofunctional acrylate ester, monofunctional methacrylate ester, substituted derivatives of the foregoing, and blends of the foregoing. Methacrylate esters are especially useful, particularly desirable examples of which include methyl methacrylate, ethyl methacrylate, butyl methacrylate, methoxy ethyl methacrylate, cyclohexyl methacrylate, tetrahydrofurfuryl methacrylate, and blends thereof. Blends of alkyl methacrylate (e.g., methyl methacrylate) and alkyl acrylate (especially those in which the alkyl group has from 4 to 10 carbon atoms, e.g., butyl acrylate) are also quite useful.

Acrylic compositions of the invention are uniquely useful in providing adhesives, and adhesive compositions of the invention advantageously provide reduced levels of mobile constituents, excellent solvent resistance, and adhesion to low surface energy plastics such as polyethylene, polypropylene, polytetrafluoroethylene, and the like. The term "low energy surface" means those materials having a surface energy of less than 45 mJ/m^2 , more typically less than about $35\text{-}40 \text{ mJ/m}^2$.

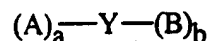
In another aspect, the invention relates to a method of initiating the polymerization of an acrylic

monomer, the method comprising the steps of providing at least one acrylic monomer, blending the at least one acrylic monomer with the components of the polymerization initiator system according to the invention, and initiating polymerization of the at least one acrylic monomer. When the organoborane amine complex of the initiator system comprises a polyamine, this method may be used to prepare a polymerizable acrylic composition having improved solvent resistance.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

Decomplexer

The decomplexer of the invention preferably comprises at least one free radically polymerizable group and at least one amine-reactive group in the same molecule. Examples of useful decomplexers of the invention include the following:

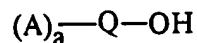


wherein "A" is a group that is capable of forming a covalent bond with acrylic monomer; "B" is a group that is capable of forming a covalent bond with the amine portion of the organoborane amine complex; "Y" is a polyvalent organic linking group; "a" represents the number of free radically polymerizable groups; and "b" represents the number of amine-reactive groups.

Group "A" preferably comprises free radically polymerizable such as an alkene group. The alkene group may be unsubstituted or substituted or part of a cyclic ring structure. Substituted alkenes include, for example, those alkenes having alkyl or aryl group substitution. Preferred alkenes are those having terminal unsubstituted double bonds such as allyl groups. Even more preferred alkenes are styryls, while the most preferred alkenes are acrylics.

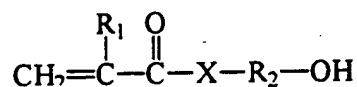
Amine-reactive group "B" preferably comprises an isocyanate group. Because the decomplexer comprises at least one free radically polymerizable group and at least one amine-reactive group, the value of each of "a" and "b" is at least one. Preferably, the sum of "a" and "b" is less than or equal to six, more preferably less than or equal to four, most preferably two.

Polyvalent organic linking group "Y" may comprise a wide variety of different chemical structures depending on the reagents used to prepare the decomplexer. Preferably, the decomplexer comprises the reaction product of a hydroxyl compound containing a free radically polymerizable group and a polyisocyanate. Useful hydroxyl compounds include those having the structure:



wherein "A" and "a" are defined above and Q is a polyvalent organic linking group such as, for example, an alkyl, aryl, cycloalkyl, alkylene ether or carbonyl group, or combinations thereof.

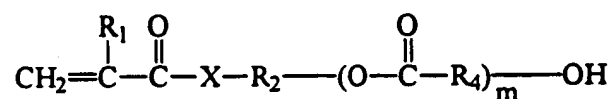
Preferred hydroxyl compounds include hydroxylated (meth)acrylates and (meth)acrylamides represented by the following structure:



wherein R_1 is hydrogen, methyl, ethyl or chlorine; R_2 is an alkyl group preferably having 2 to 16 carbon atoms; and X is oxygen or NR_3 where R_3 is hydrogen or an alkyl group preferably having 1 to 4 carbon atoms. More

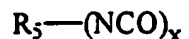
preferably, R_1 is hydrogen or methyl, R_2 is an alkyl group having 2 to 4 carbon atoms, and X is oxygen.

Adducts of the above-mentioned hydroxylated (meth)acrylates or (meth)acrylamides with lactones are also particularly useful. The adducts, hydroxyacrylate polyesters, may be represented by the following structure:



wherein R_1 , R_2 and X are defined above, R_4 is an alkyl group preferably having 2 to 11 carbon atoms and m is an integer greater than or equal to 1. More preferably, R_1 is hydrogen or methyl, R_2 is an alkyl group having 2 to 4 carbon atoms, X is oxygen, R_4 is an alkyl group having 3 to 5 carbon atoms, and m is an integer from 1 to 4.

Useful polyisocyanates may be represented by the structure:



wherein R_5 is an alkyl, aryl, or an arylalkyl group and x is at least 2, more preferably between 2 and 4, inclusively. Examples of useful aromatic isocyanates include 1,4-phenylene diisocyanate, 2,4- and 2,6-toluene diisocyanate, diphenylmethane diisocyanate, 1,5-naphthalene diisocyanate, and triphenylmethane triisocyanate. Examples of useful aliphatic diisocyanates include 1,6-hexamethylene diisocyanate, 2,2,4-trimethylhexane diisocyanate, 1,4-cyclohexyl diisocyanate, isophorone diisocyanate, dicyclohexylmethane diisocyanate, and meta-tetramethylxylylene diisocyanate.

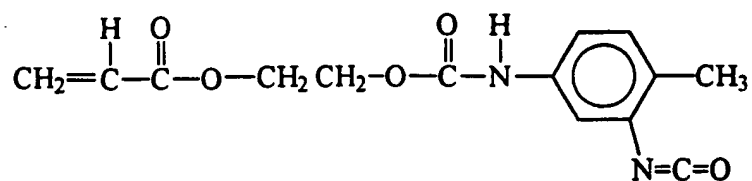
Other useful isocyanates include the oligomeric reaction products of any of the above aliphatic and

aromatic isocyanates with each other. Examples include dimers having carbodiimide or uretdione linkage or trimers having biuret or isocyanurate linkages.

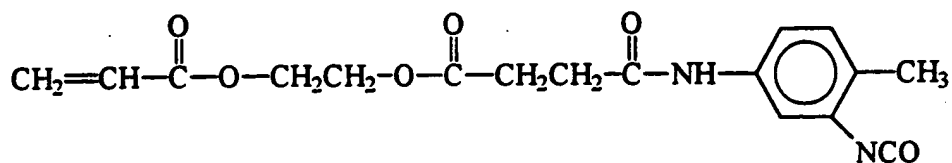
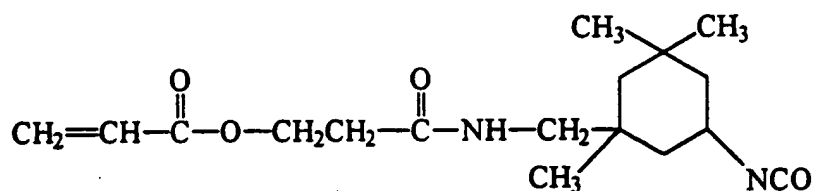
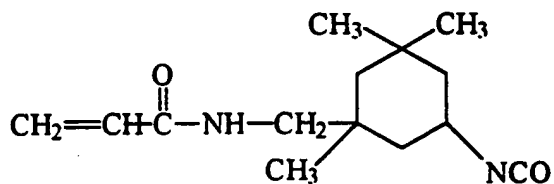
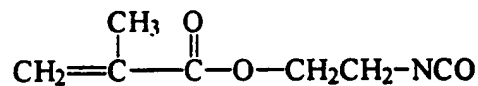
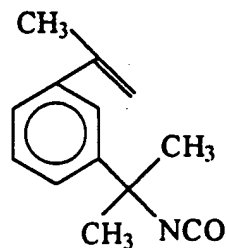
-Adducts prepared by reacting a molar excess of the above polyisocyanates with active hydrogen containing compounds are also useful to make the decomplexer of the invention. Examples of useful active hydrogen containing compounds include polyols, polythiols, and polyamines. Examples of useful polyols include alkylene glycols such as 1,2-ethanediol, 1,4-butanediol, and 1,6-hexanediol; alkylene ether glycols such as diethylene glycol, tripropylene glycol, and poly(butylene glycol); polyhydroxy alkanes such as 4-cyclohexane dimethanol, glycerine, trimethylolpropane, and pentaerythritol. Examples of useful polythiols include 1,3-propanedithiol, 2,2'-dimercapto diethyl ether, 2,2'-dimercapto diethyl sulfide, triethylene glycol dimercaptan, and trimethylolethane tri(3-mercaptopropionate). Examples of useful polyamines include ethylenediamine, 1,3-diaminopropane, 1,6-hexanediamine and 4,7,10-trioxa-1,13-tridecanediamine.

The most preferred decomplexers of the invention comprise the reaction product of isophorone diisocyanate (IPDI) with hydroxyethylmethacrylate (HEMA), hydroxyethylacrylate (HEA), or hydroxybutylacrylate (HBA); IPDI or toluene diisocyanate (TDI) with either the adduct of HEA and caprolactone or HEMA and caprolactone, and TDI with HEA.

For example, the decomplexer comprising the reaction product of TDI and HEA may be represented as follows:



Other examples of decomplexers of the present invention include the following:



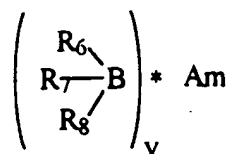
As stated previously, the decomplexer of the invention comprises at least one free radically

polymerizable group and at least one isocyanate group in the same molecule. The total number of each type of group depends on the molar ratio and chemical structure of the starting reagents used to make the decomplexer. For example, when a triisocyanate is reacted with one mole of a hydroxylated (meth)acrylate, the resulting decomplexer is expected to have two isocyanate groups and one free radically polymerizable group. By adjusting the molar ratio, however, it is possible to obtain one isocyanate group and two free radically polymerizable groups.

The decomplexer of the invention may be prepared by allowing the starting reagents (for example, hydroxylated (meth)acrylate and diisocyanate) to react in the presence of a catalyst (such as dibutyltindilaurate) that creates urethane linkages and optionally a free radical inhibitor (typically a phenolic compound such as hydroquinone monomethylether or butylated hydroxy toluene). When the starting reagents are a mono-hydroxylated (meth)acrylate and diisocyanate, the molar ratio of the former to the latter is preferably 0.9-1.1:1; more preferably an equimolar quantity of each is used to provide an NCO to OH equivalent ratio of 2:1. The catalyst is typically provided at a level of 0.01 weight percent based on the total amount of decomplexer prepared. The free radical inhibitor (if used) is typically provided at a level of 600 parts by weight per million parts by weight decomplexer. Additional protection against premature free radical polymerization may be provided by bubbling dry air through the co-reactants. The reaction is conveniently carried out at room temperature.

Organoborane Amine Complex

In general, complexes of the invention are complexes of organoborane and amine. Such complexes may be represented by the structure:

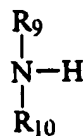


R_6 is an alkyl group having 1 to 10 carbon atoms. R_7 and R_8 are independently selected from alkyl groups having 1 to 10 carbon atoms and phenyl-containing groups. Preferably, R_6 , R_7 and R_8 are independently selected alkyl groups having 1 to 5 carbon atoms. Most preferably, they are the same.

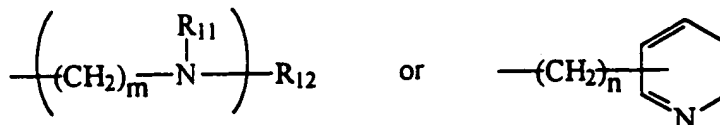
The value of v is selected so as to provide an effective ratio of primary or secondary amine nitrogen atoms to boron atoms in the complex. The ratio of primary or secondary amine nitrogen atoms to boron atoms in the complex should broadly be about 0.5:1 to 4:1.

The amine, Am, may comprise a variety of structures, for example, any primary or secondary amine, polyamine containing a primary or secondary amine, or ammonia. When solvent resistance in the ultimate polymerized acrylic composition is desired, Am preferably comprises a polyamine.

For example, Am may be represented by the structure:

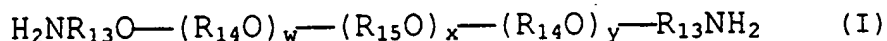


where R_9 and R_{10} are hydrogen, alkyl of 1 to 10 carbon atoms, ethanol or

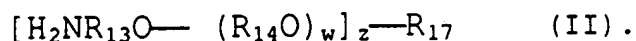


where R_{11} and R_{12} are hydrogen or alkyl of 1 to 10 carbon atoms, m is greater than 2, and n is 1 to 6. The alkyl groups in these formulae may comprise straight or branched chains. Illustrative compounds include *n*-octylamine, 1,6-diaminohexane, diethylamine, dibutylamine, dipropylenediamine, diethylenetriamine, ammonia, 1,3-propylenediamine, and 1,2-propylenediamine.

Alternatively, A_m may be a polyoxyalkylenepolyamine having a structure such as



or



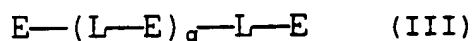
R_{13} , R_{14} and R_{15} are preferably alkylene groups having 1 to 10 carbon atoms which may be the same or different. More preferably, R_{13} is an alkyl group having 2 to 4 carbon atoms, R_{14} is an alkyl group of 2 or 3 carbon atoms, and R_{15} is an alkyl group of 2 or 3 carbon atoms. R_{17} is a residue of a polyol used to prepare the polyoxyalkylenepolyamine (i.e., the organic structure that remains when the hydroxyl groups are removed). R_{17} may be branched or linear, saturated or unsaturated, and substituted or unsubstituted.

The value of w is ≥ 1 (more preferably about 1 to 150, and most preferably about 1 to 20). The value of x and y are ≥ 0 . The value of z is > 2 (more preferably 3 or 4).

Examples of useful polyoxyalkylenepolyamines include polyethyleneoxidediamine, polypropyleneoxidediamine, diethyleneglycolpropylenediamine, triethyleneglycolpropylenediamine, polytetramethyleneoxidediamine, and polyethyleneoxide-co-polypropyleneoxidediamine.

Examples of useful commercially available polyoxyalkylenepolyamines include various JEFFAMINES from Huntsman Chemical Company such as the D, ED, EDR, and T series diamines (e.g., D-400, D-2000, D-5000, ED-600, ED-900, ED-2001, EDR-148 and T403), as well as DCA-221 from Dixie Chemical Company.

Alternatively, the amine, Am, may comprise the reaction product of one or more diprimary amine-terminated materials (i.e., the two terminal groups are primary amine) and one or more materials containing at least two groups reactive with the primary amine (the latter may be referred to as a "difunctional primary amine-reactive" material). In this case, Am may be represented by the structure:



in which each E group is the residue of the diprimary amine-terminated material and each L is a linking group that is the residue of the material having at least two groups reactive with primary amine. The term "residue" means those portions of the diprimary amine-terminated material and the difunctional primary amine-reactive material that remain after reaction to form the polyamine adduct. The integral value of q is ≥ 0 (more preferably, it is from 0 to 5, and most preferably it is 0 or 1).

Useful diprimary amine-terminated materials (E) include alkyl diprimary amines, aryl diprimary amines, alkaryl diprimary amines, polyoxyalkylenediamines such as those represented by structure I described above, and mixtures thereof. A particularly preferred material for E is an aliphatic oxyalkylene diamine sold as DCA-221 from Dixie Chemical Company.

Suitable candidates for the material having at least two groups reactive with primary amine (L) may be represented by the general structure W-R-Z, wherein W

and Z are moieties independently selected from the group consisting of carboxylic acid, carboxylic acid halide, ester, aldehyde, epoxide, amino alcohol, and acrylic, and R is a divalent organic radical. Most preferably W and Z are the same. A particularly preferred material for L is hexanedioldiacrylate.

Initiator System

The initiator systems of the invention are especially useful in polymerizing acrylic monomers, particularly for making polymerizable acrylic adhesives. The organoborane amine complex is employed in an effective amount, which is an amount large enough to permit polymerization to readily occur to obtain a polymer (preferably, an acrylic polymer) of high enough molecular weight for the desired end use. If the amount of organoborane amine complex is too low, then the polymerization may be incomplete or, in the case of adhesives, the resulting composition may have poor adhesion. On the other hand, if the amount of organoborane amine complex is too high, then the polymerization may proceed too rapidly to allow for effective mixing and use of the resulting composition. Large amounts of complex could also lead to the generation of large volumes of borane, which, in the case of an adhesive, could weaken the bondline. The useful rate of polymerization will depend in part on the method of applying the composition to a substrate. Thus, a faster rate of polymerization may be accommodated by using a high speed automated industrial adhesive applicator rather than by applying the composition with a hand applicator or by manually mixing the composition.

Within these parameters, an effective amount of the organoborane amine complex is an amount that preferably provides about 0.03 to 1.5 weight % boron,

based on the total weight of the adhesive composition, more preferably about 0.04 to 0.60 weight % boron.

The weight % of boron in a composition is equal to the following:

$$\frac{(\text{weight of complex in the composition})}{(\text{Total weight of composition})} \times (\text{weight \% of boron in the complex})$$

The organoborane decomplexer liberates organoborane by reacting with the amine, thereby removing the organoborane from chemical attachment with the amine.

The decomplexer is employed in an effective amount; that is, an amount effective to promote polymerization by liberating organoborane from the complex but without materially adversely affecting the properties of the ultimate polymerized composition. Larger amounts of decomplexer may permit the polymerization to proceed too quickly and, in the case of adhesives, the resulting materials may demonstrate inadequate adhesion to low energy surfaces. If small amounts of decomplexer are employed, the rate of polymerization may be too slow and the monomers that are being polymerized may not adequately increase in molecular weight. However, a reduced amount of decomplexer may be helpful in slowing the rate of polymerization if it is otherwise too fast.

Within these parameters, the decomplexer may be provided in an amount wherein the number of equivalents of amine reactive groups is as much as twice stoichiometric with the number of amine groups in the organoborane amine complex. The number of amine groups includes both primary and secondary amine groups if the amine is a polyamine. However, it is much more preferred that the number of equivalents of amine reactive groups be stoichiometric with the total number of amine groups in the organoborane amine complex.

Compositions

The initiator systems of the invention are useful in providing polymerizable acrylic monomer compositions. "Acrylic monomers" are polymerizable monomers having one or more acrylic or substituted acrylic moieties, chemical groups or functionality; that is, groups having the general structure

$$\begin{array}{c} R_{18} \quad O \\ | \quad || \\ H_2C=C-C-X-R_{19} \end{array}$$

wherein R_{18} and R_{19} are hydrogen or organic radicals, and X is oxygen or NR_3 where R_3 is hydrogen or an alkyl group preferably having 1 to 4 carbon atoms. Where R_{18} and R_{19} are organic radicals, they may be the same or they may be different. Blends of acrylic monomers may also be used. The polymerizable acrylic monomer may be monofunctional, polyfunctional or a combination thereof.

The most useful monomers are monofunctional acrylate and methacrylate esters or amides and substituted derivatives thereof such as cyano, chloro, and silane derivatives as well as blends of substituted and unsubstituted monofunctional acrylate and methacrylate esters and amides. Particularly preferred monomers include lower molecular weight methacrylate esters and amides such as methyl methacrylate, ethyl methacrylate, butyl methacrylate, methoxy ethyl methacrylate, cyclohexyl methacrylate, tetrahydrofurfuryl methacrylate, N,N-dimethyl methacrylamide and blends thereof.

Acrylate esters and amides and higher molecular weight methacrylate esters and amides are less preferred for use alone, but can be especially usefully employed as modifying monomers with predominating amounts of lower molecular weight methacrylate esters and amides so as to, for example, enhance the softness or flexibility of the ultimate composition. Examples

of such acrylate esters and amides and higher molecular weight methacrylate esters and amides include methyl acrylate, ethyl acrylate, isobornyl methacrylate, butyl acrylate, n-octyl acrylate, 2-ethylhexyl acrylate, 2-ethylhexyl methacrylate, decylmethacrylate, dodecyl methacrylate, tert-butyl methacrylate, acrylamide, N-methyl acrylamide, diacetone acrylamide, N-tert-butyl acrylamide, N-tert-octyl acrylamide, N-decyl methacrylamide, gamma-methacryloxypropyl trimethoxysilane, 2-cyanoethyl acrylate, 3-cyanopropyl acrylate, tetrahydrofurfuryl chloroacrylate, glycidyl acrylate, glycidyl methacrylate, and the like. Dimethylaminoethyl acrylate and dimethylaminoethyl methacrylate may also be used as modifying agents although additional organoborane liberator may be required due to the extra amine groups.

Particularly preferred are blends of any of the lower molecular weight alkyl methacrylate esters described above with alkyl acrylates having 4 to 10 carbon atoms in the alkyl group, such as blends of methyl methacrylate and butylacrylate. Polymerizable compositions of this type may broadly comprise, based on the total weight of the composition, about 2 to 50 wt. % of the alkyl acrylate and, correspondingly, about 50 to 98 wt. % of the alkyl methacrylate.

Other useful monomers that can be used include di- or higher functional (meth)acrylates (for example, hexanedioldiacrylate, ethylene glycol dimethacrylate, ethylene glycol diacrylate, triethylene glycol dimethacrylate, polyethylene glycol diacrylate, tetraethylene glycol di(meth)acrylate, glycerol diacrylate, diethylene glycol dimethacrylate, pentaerythritol triacrylate, trimethylolpropane trimethacrylate, as well as other polyether diacrylates and dimethacrylates).

The above-discussion of acrylic monomers is not intended to be exhaustive. Other acrylic monomers are known in the art and may be used in the present invention.

The compositions may further comprise a variety of optional additives. One particularly useful additive is a thickener such as medium (about 100,000) molecular weight polymethyl methacrylate which may be incorporated in an amount of about 10 to 40 weight %, based on the total weight of the composition. Thickeners may be employed to increase the viscosity of the composition to a more easily applied viscous syrup-like consistency.

Another particularly useful additive is an elastomeric material. These materials can improve the fracture toughness of compositions made therewith which can be beneficial when, for example, bonding stiff, high yield strength materials such as metal substrates that do not mechanically absorb energy as easily as other materials, such as flexible polymeric substrates. Such additives can be incorporated in an amount of about 5% to 35% by weight, based on the total weight of the composition.

Useful elastomeric modifiers include chlorinated or chlorosulphonated polyethylenes such as HYPALON 30 (commercially available from E. I. duPont de Nemours and Co., Wilmington DE). Also useful, and even more preferred, are certain graft copolymer resins such as particles that comprise rubber or rubber-like cores or networks that are surrounded by relatively hard shells, these materials often being referred to as "core-shell" polymers. Most preferred are the acrylonitrile-butadiene-styrene graft copolymers.

In addition to improving the fracture toughness of the composition, core-shell polymers can also impart enhanced spreading and flow properties to the uncured

composition. These enhanced properties may be manifested by a reduced tendency for the composition to leave an undesirable "string" upon dispensing from a syringe-type applicator, or sag or slump after having been applied to a vertical surface. Use of more than about 20% of a core-shell polymer additive is desirable for achieving improved sag-slump resistance.

Peroxides may be optionally included (typically in an amount of about 2 % by weight or less, based on the total weight of the composition), for example, to adjust the speed at which the compositions polymerize or to complete the polymerization.

Small amounts of inhibitors such as hydroquinone monomethylether may be used, for example, to prevent or reduce degradation of the acrylic monomers during storage. Inhibitors may be added in an amount that does not materially reduce the rate of polymerization or the ultimate properties of an adhesive or other composition made therewith, typically about 100 - 10,000 ppm based on the weight of the polymerizable monomers.

Other possible additives include non-reactive colorants, fillers (e.g., carbon black), etc. provided they are dried sufficiently to reduce moisture content below about 0.1 % by weight based on the total weight of the additives.

The various optional additives are employed in an amount that does not significantly adversely affect the polymerization process or the desired properties of compositions made therewith.

Polymerizable acrylic compositions according to the invention may be used in a wide variety of ways, including as sealants, coatings, and injection molding resins. They may also be used as matrix resins in conjunction with glass and metal fiber mats such as in resin transfer molding operations. They may further be

used as encapsulants and potting compounds such as in the manufacture of electrical components, printed circuit boards and the like. Quite desirably, they provide polymerizable acrylic adhesive compositions that can bond a diverse myriad of substrates, including polymers, wood, ceramics, concrete, and primed metals.

The polymerizable compositions of the invention can be easily used as two-part adhesives. The components of the polymerizable composition are blended as would normally be done when working with such materials. The decomplexer of the polymerization initiator system is usually included in this blend so as to separate it from the organoborane amine complex, thus providing one part of the two-part composition. The organoborane amine complex of the polymerization initiator system provides the second part of the composition and is added to the first part shortly before it is desired to use the composition. The complex may be added to the first part directly or it may be predissolved in an appropriate carrier such as a small amount of methyl methacrylate.

For a two-part adhesive such as those of the invention to be most easily used in commercial and industrial environments, the ratio at which the two parts are combined should be a convenient whole number. This facilitates application of the adhesive with conventional, commercially available dispensers. Such dispensers are shown in U.S. Patent Nos. 4,538,920 and 5,082,147 and are available from Conprotec, Inc. (Salem NH) under the tradename "Mixpac."

Once the two parts have been combined, the composition should be used quickly, as the useful pot life may be short depending upon the acrylic monomer mix, the amount of complex, and the temperature at which the bonding is to be performed.

The polymerizable composition is applied to one or both substrates and then the substrates are joined together with pressure to force excess composition out of the bond line. This also has the advantage of displacing composition that has been exposed to air and that may have advanced too far in cure. In general, the bonds should be made shortly after the composition has been applied, preferably within about 10 minutes. The typical bond line thickness is about 0.1 to 0.3 mm but may exceed 1.0 mm when gap filling is needed. The bonding process can easily be carried out at room temperature and to improve the degree of polymerization it is desirable to keep the temperature below about 40°C, preferably below 30°C, and most preferably below about 25°C.

The bonds will cure to a reasonable green strength to permit handling of the bonded components within about 1 hour. Full strength will be reached in about 24 hours under ambient conditions; post-curing with heat (typically about 80°C) may be used if desired.

The invention will be more fully appreciated with reference to the following nonlimiting examples in which (unless noted otherwise) all weights are given as weight percents (weight %), based on the total weight of the composition which is 100 weight %.

EXAMPLES

Bond Strength Test Method

The adhesive compositions described below were tested for bond strength. The base component of a given adhesive formulation (comprising a thorough mixture of thickened monomers and decomplexer) was combined with the organoborane polyamine complex (referred to as the accelerator component) and mixed with a spatula for one minute. The adhesive was then applied to a 2 inch (5.1 cm) wide, 6 inch long (15.2

cm), 8 mil (0.2 mm) thick film. The film was held flat using two 30 mil (0.8 mm) thick aluminum overlays, the overlays overlapping 0.5 inch (1.3 cm) along the entire length of each side of the film.

Each adhesive formulation was tested on two different film substrates, skived polytetrafluoroethylene (PTFE) and skived ultrahigh molecular weight polyethylene (PE). Neither of the films had any surface treatment.

Once the adhesive was applied to the first film, a second film comprising the same substrate was eased onto the adhesive using a 2.7 lb (1.2 kg) weighted glass plate. One side of the glass plate was applied to one side of the second film at an angle of about 30 degrees and then gradually lowered to a horizontal position so that the second film was applied incrementally to the first to avoid formation of bubbles in the adhesive. Sufficient adhesive was applied to the first film to provide minimum bond dimensions of 1 inch (2.5 cm) in width by 1.5 inch (3.8 cm) in length.

The bonded adherends were allowed to cure for 24 hours at room temperature (21-25°C). The samples were then tested for bond strength in the T-peel mode using an Instron Tensile Tester set at a crosshead speed of 4 inches/min (10.2 cm/min). Two overlapping free ends of the sample were clamped into the jaws of the Instron tester, one free end in the upper jaw and one free end in the lower jaw. The jaws were then pulled apart until at least 1 inch (2.5 cm) of adhesive was exposed. The average force per width during the run after the adhesive was initially exposed was recorded as the Peel Strength in pounds/inch width (piw).

The tensile strengths of PTFE and PE films were found to be 11 pounds/inch width (20 N/cm width) and 25 pound/inch width (44 N/cm width), respectively.

Preferably, the Peel Strength values approximate the tensile strength values, indicating substrate as opposed to adhesive failure.

Solvent Resistance

2-3 gram remnants of each adhesive formulation (leftover from the above-test) were sealed in their 20 cm³ glass mixing vials for 24 hours at room temperature (21-25°C) after the bonds described above were prepared. The vial plus adhesive was then weighed to give the value denoted below as W1. The contents of each vial were then soaked in tetrahydrofuran (THF) for at least 24 hours until sufficiently swollen to be removed. When necessary, a spatula was used to break up the mass to facilitate removal. The mass of each vial was transferred into its own 4 ounce (118 cm³) glass jar, and the empty vial was dried and weighed to give the value denoted below as W2. Additional THF was added to each glass jar to bring the solids content to 5% by weight. The jar was capped and placed on a roll mixer for at least 24 hours. The fluid mixture was then poured into a medium mesh conical gauze/paper paint filter (available from Foremost Co. and having a weight denoted below as W3) and thoroughly rinsed with THF. Filters were dried in an 80°C forced air oven for one hour.

The filters were allowed to equilibrate overnight at room temperature before being weighed to give the value denoted below as W4. Adhesive percent solids was then calculated as follows:

$$\text{Percent Solids} = \frac{W4 - W3 + 0.0174}{W1 - W2} \times 100$$

It is noted that a correction factor of 0.0174 milligrams was added to W4-W3 in the equation above to

account for that portion of the filter that was lost by virtue of the process conditions.

Percent Gel was then calculated for samples of the invention by dividing the Percent Solids (calculated as described above) by the weight percent of the adhesive which is capable of addition polymerization (i.e., the portion derived from the acrylic monomers, the amine part of the complex, and the decomplexer). A sample calculation is provided below for Example 1 of the invention:

$$\text{Percent Gel} = \frac{\text{Percent Solids} \times 100}{\text{Percent Capable of Add'n Polymerization}}$$

$$(\text{Percent Gel})_{\text{Ex1}} = \frac{(\text{Percent Solids})_{\text{Ex1}} \times 100}{(\text{Wt.}\%_{\text{MMA}} + \text{Wt.}\%_{\text{BA}} + \text{Wt.}\%_{\text{DC}} + \text{Wt.}\%_{\text{AmC}})}$$

where $\text{Wt.}\%_{\text{MMA}}$ is the weight percent of methyl methacrylate, $\text{Wt.}\%_{\text{BA}}$ is the weight percent of butyl acrylate, $\text{Wt.}\%_{\text{DC}}$ is the weight percent of decomplexer, and $\text{Wt.}\%_{\text{AmC}}$ is the weight percent of the amine portion of the complexer used in Example 1. The Percent Solids of Example 1 can be found from Table V to be 68. Example 1 comprised Initiator In-1 given in Table III. The organoborane portion (triethylborane) has a molecular weight of 98, while the amine portion (1,6-hexanediamine) has a molecular weight of 116. Table IV indicates that the weight percent of Initiator was 2.510%. Thus, the value of $\text{Wt.}\%_{\text{AmC}}$ can be calculated as follows:

$$\text{Wt.}\%_{\text{AmC}} = (2.510) \frac{(116)}{(312)}$$

$$\text{Wt.}\%_{\text{AmC}} = 0.933$$

The $(\text{Percent Gel})_{\text{Ex1}}$ can now be calculated from the other weight percents listed in Table IV as follows:

$$(\text{Percent Gel})_{\text{Ex1}} = \frac{68 \times 100}{(41.179 + 27.452 + 5.982 + 0.933)}$$

$$(\text{Percent Gel})_{\text{Ex1}} = 90\%.$$

Small deviations from table values are due to rounding of Percent Solids values.

In a similar fashion, Percent Solids and Percent Gel were calculated for cured films recovered from the samples prepared for the Bond Strength Test described.

Decomplexer Synthesis

Preparation of Aliphatic Isocyanato (Meth)acrylates

Dibutyltindilaurate (DBTDL) catalyst and 2,6-di-*t*-butyl-4-methylphenol inhibitor were charged to a 50 ml glass vial in weight ratios specified in Table I. These charges were followed with charges of isophorone diisocyanate (IPDI) and a selected hydroxylated (meth)acrylate also specified in Table I. The selected hydroxylated (meth)acrylate had been dried over a bed of 4Å molecular sieves prior to charging and the molar ratio of IPDI to selected hydroxylated (meth)acrylate was 1:1. The charges were mixed thoroughly while purging the glass vial with a stream of dry air. The vial was then sealed tightly to exclude moisture and placed in a water bath at room temperature (about 25°C). The contents of the vial were allowed to react for at least 24 hours before use. These aliphatic derivatives were designated as decomplexers A-E in Table I.

Preparation of Aromatic Isocyanato (meth)acrylates

The above procedure was modified to prepare aromatic derivatives. The DBTDL catalyst was eliminated and the reaction time was increased to 48 hours. The IPDI was replaced with 2,4-diisocyanatotoluene (Mondur TDS) available from Miles

Inc. The selected hydroxylated (meth)acrylates were neutralized prior to charging. The 2-hydroxyethylacrylate was neutralized with 0.14% orthophosphoric acid and then passed through a column of activated silica gel. The selected hydroxylated (meth)acrylates, Tone® M-100 and Tone® M-200 (available from Union Carbide) were neutralized by forming a slurry with activated silica gel and then decanting through glass wool. The aromatic derivatives have been designated as decomplexers F-H in Table I.

Table I
Bireactive Decomplexers

Decomplexer	Wt.% DBTDL	Wt.% BHT	Isocyanate		Hydroxylated (Meth)acrylate	
			Name	Wt.%	Name ¹	Wt.%
A	0.010	0.043	IPDI	65.646	HEA	34.301
B	0.010	0.050	IPDI	63.030	HEMA	36.910
C	0.010	0.040	IPDI	60.625	HBA	39.325
D	0.010	0.020	IPDI	39.490	M-100	60.480
E	0.010	0.047	IPDI	38.254	M-200	61.689
F	0.010	0.043	TDS	59.968	HEA	39.979
G	0.010	0.013	TDS	33.844	M-100	66.133
H	0.010	0.047	TDS	32.688	M-200	67.255

¹ HEA = Hydroxyethyl acrylate

HEMA = Hydroxyethyl methacrylate

HBA = Hydroxybutyl acrylate

M-100 = Tone® M-100

M-200 = Tone® M-200

Comparative Decomplexers

Decomplexers which were used for comparison with the decomplexers of the present invention are designated as decomplexers I-N in Table II below. The m-TMXDI is meta-tetramethylxylene diisocyanate available from American Cyanamide and the Isonate 2143L is a noncrystallizing version of 4,4'-diphenylmethane diisocyanate available from Dow Chemical Co.

Table II
Comparative Decomplexers

<u>Decomplexer</u>	<u>Name</u>
I	benzaldehyde
J	methacrylic acid
K	IPDI
L	m-TMXDI
M	Mondur TDS
N	Isonate 2143L

Initiators

The organoborane amine complex initiators used in the present invention are listed in Table III below. All were prepared with triethyl borane using a 1:1 boron to nitrogen ratio. The organoborane polyamine complexes may be readily prepared using known techniques. Typically, the polyamine is combined with the organoborane in an inert atmosphere with slow stirring. An exotherm is often observed and cooling of the mixture is therefore recommended. If the ingredients have a high vapor pressure, it is desirable to keep the reaction temperature below about 70° to 80°C. Once the materials have been well mixed the complex is permitted to cool to room temperature. No special storage conditions are required although it is preferred that the complex be kept in a capped vessel in a cool, dark location. Advantageously, the complexes can be prepared in the absence of organic solvents that would later have to be removed, although they could be prepared in solvent if so desired. Solvents used in the preparation of the complexes should, preferably, be ones that do not coordinate amines, for example, tetrahydrofuran or hexane.

Table III
Organoborane Amine Complex Initiators

<u>Initiator</u>	<u>Organoborane</u>	<u>Amine</u>
In-1	triethylborane	1,6-hexanediamine
In-2	triethylborane	Jeffamine T-403 triamine ¹
In-3	triethylborane	Jeffamine ED-600 diamine ¹
In-4	triethylborane	E-L-E diamine ²

- ¹ available from Huntsman Corp
- ² 2:1 molar adduct of DCA-221 diamine (Dixie Chemical Co.) and 1,6-hexanediol diacrylate

Adhesive Compositions (examples 1-20 and comparative examples 21-27)

Two part adhesive compositions were prepared according to Table IV below. The weight percent of each component based upon total composition weight is shown in this table.

The two part adhesives of this invention comprise an accelerator component and a base component. The accelerator component in all examples was solely the organoborane amine complex initiator.

The base component comprised a thickened acrylic monomer mixture to which has been added an effective amount of decomplexer. The monomer mixture was a 60:40 blend by weight of methyl methacrylate (MMA) and butyl acrylate (BA) in a 25% solution of 101,000 molecular weight poly(>95% methyl methacrylate-co-<5% ethyl acrylate) (PMMA) available from Aldrich Chemical Co.

Unless otherwise indicated, the standard formulation provided 2.5 moles of triethylborane for every 100 moles of free-radically polymerizable (meth)acrylate groups and 1.1 moles of amine-reactive groups for each mole of initiator groups.

Table IV
Adhesive Compositions

Example	Base				Decomplexer		Accelerator		Borane ⁽¹⁾
	Wt% MMA	Wt% BA	Wt% PMMA	Type	Wt%	Initiator	Wt%		
1	41.179	27.452	22.877	A	5.982	In-1	2.510	2.5	
2	41.077	27.385	22.820	B	6.214	In-1	2.504	2.5	
3	40.975	27.317	22.764	C	6.446	In-1	2.498	2.5	
4	40.292	26.861	22.384	C	6.338	In-2	4.125	2.5	
5	39.122	26.081	21.734	C	6.154	In-4	6.909	2.5	
6	39.608	26.406	22.005	D	9.567	In-1	2.414	2.5	
7	37.727	25.152	20.960	D	12.905	In-1	3.256	3.5	
8	35.151	23.434	19.528	D	17.477	In-1	4.410	5.0	
9	32.106	21.404	17.837	D	22.880	In-1	5.773	7.0	
10	28.213	18.809	15.674	D	29.787	In-1	7.517	10.0	
11	39.487	26.325	21.937	E	9.844	In-1	2.407	2.5	
12	41.531	27.688	23.073	F	5.177	In-1	2.531	2.5	
13	44.277	29.518	24.598	F	1.079	In-1	0.528	0.5	
14	39.934	26.623	22.186	G	8.823	In-1	2.434	2.5	
15	40.875	27.250	22.709	G	7.184	In-1	1.982	2.0	
16	41.850	27.900	23.25	G	5.486	In-1	1.514	1.5	
17	42.862	28.575	23.812	G	3.724	In-1	1.027	1.0	
18	43.911	29.274	24.395	G	1.897	In-1	0.523	0.5	
19	39.811	26.541	22.117	H	9.104	In-1	2.427	2.5	
20	38.252	25.502	21.251	H	8.747	In-3	6.248	2.5	
Comp 21	42.996	28.664	23.887	I	1.905	In-1	2.548	2.5	
Comp 22	42.453	28.302	23.585	J	2.137	In-1	3.523	3.4	
Comp 23	42.126	28.084	23.403	J	2.891 ⁽²⁾	In-1	3.496	3.3	
Comp 24	42.957	28.638	23.865	K	1.993	In-1	2.547	2.5	
Comp 25	42.872	28.582	23.818	L	2.186	In-1	2.542	2.5	
Comp 26	43.143	28.762	23.968	M	1.569	In-1	2.558	2.5	
Comp 27	42.704	28.470	23.725	N	2.570	In-1	2.531	2.5	

(1)

moles/100 moles (meth)acrylate vinyl groups

(2)

1.5 moles of carboxylic acid groups for each mole of initiator amine groups

The bond strengths and solvent resistance of the adhesive formulations were determined using methods described above. Results are shown in Table V.

5 Table V
Bond Strength and Solvent Resistance

Example	Bond Strength				Solvent Resistance	
	PTFE		PE		% Solids	% Gel
	piw	N/cm	piw	N/cm		
1	6	10	12	21	68	89
2	6	10	3	5	66	87
3	6	10	14	25	64	85
4	9	16	2	4	73	96
5	5	9	10	17	60	78
6	5	9	10	17	67	88
7	7	12	5	9	69	90
8	9	16	2	4	71	91
9	11	19	1	2	73	93
10	10	17	1	2	76	95
11	5	9	4	7	71	93
12	11	19	2	4	62	83
13	6	10	9	16	0	0
14	11	19	2	4	63	83
15	11 ²	19	2	4	60	80
16	9	16	2	4	11	15
17	9	16	4	7	2	2
18	6	10	16	28	0	0
19	11	19	2	4	65	85
20	11 ²	19	2	4	65	85
Comp. 21	<1	<1	10	17	0	0
Comp. 22	<1	<1	17	29	- ³	- ³
Comp. 23	5	9	15	26	0	0
Comp. 24	6	10	3	5	0	0
Comp. 25	7	12	9	16	1	1
Comp. 26	12	21	1	2	0	0
Comp. 27	15 ¹	26	4	7	2	3

¹ adherend tore

² 1 inch (2.5 cm) wide adherend

³ not determined quantitatively, but no insolubles observed (i.e., everything dissolved)

Each bond strength value in Table V is the average reading in pounds per inch width (piw) or Newtons per cm (N/cm) over the length of the bond which was pulled. In general, preferred bond strengths are at least 5 piw (9 N/cm). The data in Table V indicate good bond formation on PTFE with most formulations. With proper formulation, good bonds were obtained on PE also.

The data of Table V also demonstrate the surprising result that solvent resistance is greatly improved when the decomplexers of the present invention are used. All of the examples of the present invention (except for examples 13, 16, 17, and 18) exhibited a % solids content after extraction of greater than 60%. When this number is corrected for only that portion of the adhesive that is capable of addition polymerization (i.e., the free-radically polymerizable vinyl compounds plus the polyamine portion of the organoborane polyamine complex if its reaction product with the decomplexer creates a new free-radically polymerizable vinyl compound through covalent or ionic bonding), the % solids increased to about 80% or more as indicated by the % Gel column in Table V. In contrast, the examples comprising the comparative decomplexers of Table II were all nearly completely soluble in the THF solvent, showing poor solvent resistance.

The poor solvent resistance which was observed for examples 13, 16, 17 and 18 may have resulted from an excessive supply of oxygen during cure, a condition which can exist when curing 2-3 grams of material in a 20 cm³ vial. In order to compare the solvent resistance of adhesives cured in the bulk versus adhesives cured as sandwiched thin films, selected adhesives were tested for solvent resistance after completion of the bond strength test. This was done by peeling the thin films of cured adhesive from the adherends after the

bond strength peel tests were completed. The adhesive films were weighed and placed in a 4 ounce (118 cm³) glass jar. Sufficient tetrahydrofuran (THF) was added to the glass jar so that the contents comprised 5 parts by weight cured adhesive film and 95 parts by weight THF. The jar was capped and placed on a roll mixer for at least 24 hours. The solvent resistance of the thin films was then determined using the procedure described earlier for determining the solvent resistance for the bulk cured adhesives. Results are reported in Table VI below.

Table VI
Comparison of % Insolubles from Bulk vs Free Film

Example	Bulk		Free Film	
	% Solids	% Gel	% Solids	% Gel
12	62	83	77	102
13	0	0	58	77
16	11	15	69	90
17	2	2	63	84
Comp. 24	0	0	8	12
Comp. 26	0	0	0	0

Adhesive compositions comprising the decomplexers of the present invention showed an increase in solvent resistance when cured as free films compared to bulk cure. This supports the theory that the poor solvent resistance observed for some examples of the invention may have been due to an excessive supply of oxygen during bulk cure. Free films from adhesives prepared with the comparative decomplexers of Table II, however, did not exhibit improved solvent resistance.

It is noted that preferred % Gel values for adhesives cured as sandwiched thin films exceed 15, more preferably exceed 30, most preferably exceed 50.

Various modifications and alterations of this invention will become apparent to those skilled in the art without departing from the scope and spirit of the

invention. It should be understood that this invention is not limited to the illustrative embodiments set forth herein.

CLAIMS

The embodiments for which an exclusive property or privilege is claimed are defined as follows:

5

1. An initiator system capable of initiating the polymerization of acrylic monomers comprising (1) an organoborane amine complex and (2) a bireactive decomplexer.

10

2. An initiator system according to Claim 1 wherein said decomplexer comprises at least one free radically polymerizable group and at least one amine-reactive group in the same molecule.

15

3. An initiator system according to Claim 1 wherein said decomplexer comprises at least one alkene group and at least one isocyanate group in the same molecule.

20

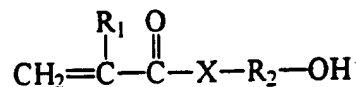
4. An initiator system according to Claim 1 wherein said decomplexer comprises at least one acrylic group and at least one isocyanate group in the same molecule.

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5. An initiator system according to Claim 1 wherein said decomplexer comprises the reaction product of a hydroxyl compound containing a group capable of forming a covalent bond with an acrylic monomer and a polyisocyanate.

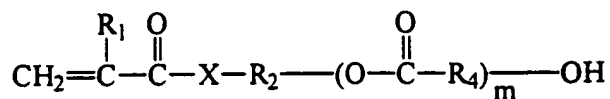
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6. An initiator system according to Claim 5 wherein said hydroxyl compound has the structure:



wherein R_1 is selected from the group consisting of hydrogen, methyl, ethyl and chlorine; R_2 is selected from the group consisting of alkyl groups; and X is selected from the group consisting of oxygen and NR_3 ,
 5 wherein R_3 is selected from the group consisting of hydrogen and alkyl groups.

7. An initiator system according to Claim 5 wherein said hydroxyl compound has the structure:



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wherein R_1 is selected from the group consisting of hydrogen, methyl, ethyl and chlorine; R_2 is selected from the group consisting of alkyl groups; X is selected from the group consisting of oxygen and NR_3 ,
 15 wherein R_3 is selected from the group consisting of hydrogen and alkyl groups; R_4 is selected from the group consisting of alkyl groups; and m is an integer greater than or equal to 1.

20 8. A polymerizable acrylic composition comprising:

- a) at least one acrylic monomer; and
- b) the initiator system of Claim 1.

25 9. A polymerized acrylic adhesive composition according to Claim 8.

30 10. A bonded composite comprising a first substrate, a second substrate and a polymerized acrylic adhesive composition according to Claim 9 that adhesively bonds the first and second substrates together.

11. A bonded composite according to Claim 10 wherein at least one of said substrates has a low energy surface.

5 12. A method of initiating the polymerization of at least one acrylic monomer, the method comprising the steps of:

- a) providing at least one acrylic monomer;
- b) blending the at least one acrylic monomer with the components of a polymerization initiator system, said system comprising:
 - 10 i) an organoborane amine complex,
 - 15 ii) an effective amount of a decomplexer comprising at least one acrylic group and at least one isocyanate group, said decomplexer being reactive with amine for liberating the organoborane to initiate polymerization of the at least one acrylic monomer; and
- 20 c) initiating polymerization of the at least one acrylic monomer.

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13. A method of increasing the solvent resistance of a polymerized acrylic adhesive composition obtained by organoborane-initiated polymerization of acrylic monomer, said method comprising the steps of

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- a) providing at least one acrylic monomer;
- b) providing a bireactive decomplexer;
- c) blending the bireactive decomplexer with the at least one acrylic monomer;

- 5
- d) providing and blending an organoborane polyamine complex to the blend formed in step c) whereby the polyamine portion of the complex reacts with and covalently bonds to the bireactive decomplexer, thereby liberating the organoborane portion of the complex; and
- 10
- e) initiating polymerization of the at least one acrylic monomer with the liberated organoborane portion of the complex and continuing polymerization until a solvent resistant acrylic adhesive composition is obtained.

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 C09J4/00 C08F4/52

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 C09J C09D C08F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP,A,0 511 464 (NATIONAL STARCH AND CHEMICAL INVESTMENT HOLDING CORPORATION) 4 November 1992 cited in the application see page 3, line 46 see page 5, line 47 - line 49 & US,A,5 106 928	1-13
P,X	WO,A,95 22567 (MINNESOTA MINING & MANUFACTURING COMPANY) 24 August 1995 see page 7, line 17 - line 19 see page 30, line 25-26 see page 17, line 10 - line 29	1-13
X	EP,A,0 051 796 (HENKEL) 19 May 1982 see page 9, line 9 - line 14	1-13

☐ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

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Date of the actual completion of the international search

17 September 1996

Date of mailing of the international search report

- 3. 10. 96

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INTERNATIONAL SEARCH REPORT

International Application No.
PCT/US 96/10341

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-511464	04-11-92	US-A- 5106928	21-04-92
		DE-D- 69201093	16-02-95
		DE-T- 69201093	11-05-95
		ES-T- 2067965	01-04-95
		HK-A- 51195	13-04-95
		JP-A- 5098216	20-04-93
		JP-B- 7072264	02-08-95
		KR-B- 9509549	24-08-95
		US-A- 5143884	01-09-92

WO-A-9522567	24-08-95	AU-A- 2971695	04-09-95
		US-A- 5539070	23-07-96

EP-A-51796	19-05-82	DE-A- 3041904	13-05-82
		AT-T- 4910	15-10-83
		JP-C- 1625570	18-11-91
		JP-B- 2050922	05-11-90
		JP-A- 57108102	06-07-82
		US-A- 4385153	24-05-83
